## In-situ STEM Observation of Strain Field Movement in a LiMn<sub>2</sub>O<sub>4</sub> Nanowire Battery

Soyeon Lee<sup>1,2</sup>, Yoshifumi Oshima<sup>2,3</sup>, Eiji Hosono<sup>4</sup>, Haoshen Zhou<sup>4</sup>, Ryoji Kanno<sup>5</sup> and Kunio Takayanagi<sup>2,6</sup>

<sup>1.</sup> Quantum Nanoelectronics Research Center, Tokyo Institute of Technology, 2-12-1-H-51 Ohokayama, Meguro-ku, Tokyo 152-8551, Japan.

<sup>2</sup> JST–CREST, 7-gobancho, Chiyoda-ku, Tokyo 102-0075, Japan.

<sup>3.</sup> School of Materials Science, Japan Advanced Institute of Science and Technology, 1-1 Asahidai, Nomi, 923-1292, Japan.

<sup>4.</sup> Energy Technology Research Institute, National Institute of Advanced Industrial Science and Technology, Umezono, 1-1-1, Tsukuba, 305-8568, Japan.

<sup>5</sup> Department of Electronic Chemistry, Tokyo Institute of Technology, G1-1 4259 Nagatsuta, Midori-ku, Yokohama 226-8502, Japan.

<sup>6</sup> Department of Physics, Tokyo Institute of Technology, 2-12-1-H-51 Oh-okayama, Meguro-ku, Tokyo 152-8551, Japan.

To understand lithium ion transport behavior is an issue for improving the performance of lithium ion batteries. Lithium insertion or extraction reaction in most lithium ion battery (LIB) electrode materials causes lattice parameter change. It leads local strain field change. Thus, measuring strain field during charge-discharge cycles allows us to know how lithium ions move inside a LIB material.

There are several methods to measure local strain field [1-3]. Among them, scanning moiré fringe (SMF) method is reported to use interference between scan raster of scanning transmission electron microscope (STEM) probe and a lattice plane of a crystal [4]. SMF method has some advantages: SMF image can be taken easily by controlling scan raster distance and direction, and strain field movement can be detected by controlling scan speed.

In this study, we have demonstrated in-situ observation of strain field movement inside  $LiMn_2O_4$  cathode during a charge-discharge cycle by using SMF method. The in-situ observation was performed by using our developed nanowire battery which consists of nanowire  $LiMn_2O_4$  cathode, ionic liquid electrolyte and  $Li_4Ti_5O_{12}$  anode [5-6]. The nanowire-battery was loaded in our homemade electrical biasing double-tilt TEM holder. The  $LiMn_2O_4$  nanowire in the nanowire-battery was observed by annular bright field (ABF) imaging [7-8] of STEM using an aberration corrected TEM, R005 at 300 kV. The convergent semi-angle was 30 mrad. The inner-outer semi-angle of ABF detector was 15-30 mrad. Simultaneously electrochemical properties were measured by cyclic voltammetry. The voltage was scanned from 2.20 to 4.50 V vs  $Li/Li^+$  at scan rate of 0.55 mV/s. The measurement was performed by source-measurement unit, Keithley 2635A.

The SMF image was taken by tilting the scan direction slightly from the (111) lattice plane of the  $LiMn_2O_4$  nanowire in order to enhance the sensitivity of the strain field (Fig. 1): the strain change of about 0.1 % was detected. As shown in Fig. 1, the moiré fringe is rotated when the spacing of the lattice fringe is expanded or shrunk.

The moiré fringe was straight without charging or discharging the nanowire battery. However, we found

that the moiré fringe was bent into the shape of mirrored "S" when lithium extraction or insertion reaction occurred. The "S" moiré indicates that the lattice spacing is gradually expanded and shrunk. Such an "S" moiré was observed discretely in the series of STEM images, indicating that the strain field was not caused by the nanowire own structure. Since upper-half and lower-half of "S" moiré was observed in sequentially obtained two STEM images, it indicates that the strain field moved along the nanowire during a charge-discharge cycle. From the series of STEM images, the estimated velocity of the strain field moving was same order with the velocity to lithium movement estimated from simultaneously obtained cyclic voltammetry curve. Therefore, we conclude that the strain field movement corresponded to the movement of lithium ions due to charging or discharging. From the analysis of SMF images, the detail of lithium transport behaviour will be discussed [9].

## References:

[1] M. J. Hytch, et al, Ultramicroscopy, 74 (1998), p. 131.

- [2] K. Tsuda, et al, J. Electron Microsc., 56 (2007), p. 57.
- [3] M. J. Hytch, et al, Nature, 453 (2008), p. 1086.
- [4] S. Kim, et al, Appl. Phys. Lett. 103 (2013), p. 033523.
- [5] S. Lee, et al, J. Phys. Chem. C, 117 (2013), p. 24236.
- [6] S. Lee, et al, ACS nano, 9 (2015), p. 626.
- [7] S. Lee, et al, J. Appl. Phys. 109 (2011), p. 113530.
- [8] S. Lee, et al, Jpn. J. Appl. Phys. 51 (2012), p. 020202.

[9] This work was supported by the Japan Science and Technology Agency (JST) under the CREST project.



**Figure 1.** Schematic illustration of scanning moiré fringe method used in this study (rotation moiré). Scan raster (blue) is slightly tilted from lattice plane (green). When lattice spacing is changed (case 1 and case 2), the angle and spacing of moiré fringes (as shown by red arrows) are changed largely.