Sunflower Protein Precipitation in a Tubular Precipitator

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Isoelectric precipitation of sunflower protein was carried out in a 20 m long, 6 mm internal diameter glass tubular precipitator. Effects of feed flow rate, protein concentration in the feed stream, and mean residence time on particle size distribution (PSD) were studied. The population balance equation (PBE) for the tubular precipitator was solved using the orthogonal collocation multiple shooting method. A simulated annealing method was used to determine the precipitation kinetic parameters for the nucleation rate, growth rate, aggregation rate, and breakage rate from the experimental data. Due to the lack of experimental data on the kinetics of sunflower protein precipitation in the literature, the extracted kinetics from our experiments were used to evaluate the predictive capability of the present model for experiments whose results had not been used to derive the kinetics. The model predictions of the particle size distribution along the tubular precipitator showed better agreement with the experimental data for large particles in comparison to the small particles (< 10 μm).

On a effectué la précipitation isoélectrique de la protéine de tournesol dans un précipitateur tubulaire en verre de 20 m de long et 6 mm de diamètre intérieur. Les effets du débit d’alimentation, de la concentration en protéine dans le courant d’alimentation et du temps de séjour moyen sur la distribution de taille des particules (PSD) ont été étudiés. L’équation de bilan de populations (PBE) pour le précipitateur tubulaire a été résolue à l’aide d’une méthode de type collocation orthogonale à tirs multiples. On a employé une méthode de retrait simulée pour déterminer les paramètres de la cinétique de précipitation pour la vitesse de nucléation, le taux de croissance, le taux d’agrégarion et le taux de rupture à partir des données expérimentales. En raison de l’absence de données expérimentales sur la cinétique de précipitation de la protéine de tournesol dans la littérature scientifique, on a utilisé les cinétiques issues de nos expériences afin d’évaluer la capacité prédictrive du présent modèle pour des expériences dont les résultats n’ont pas été utilisés pour le calcul des cinétiques. Les prédictions du modèle de la distribution de taille des particules le long du précipitateur tubulaire montrent un meilleur accord avec les données expérimentales pour des particules larges comparativement aux petites particules (< 10 μm).

Keywords: isolectric precipitation, tubular precipitator, precipitation kinetics, population balance.

Defatted oilseed meals are known to contain large amounts of nutritional proteins, which can be recovered for use in food and pharmaceutical industries. Protein recovery from solution is often achieved through precipitation, followed by solid-liquid separation. The efficiency of the subsequent processes in solids protein recovery, i.e. sedimentation or centrifugation and spray drying, depend on the size of the particles and their strength to withstand the hydrodynamic shear. To minimize losses of the solids, it is essential to maximize the mean particle size and the particle strength and minimize the spread of particle size distribution (PSD).

Studies on isolectric precipitation of oil seeds proteins have been done by several workers using different types of precipitators. Studies on batch precipitators have been reported by: Virkar et al. (1982), Petenate and Glatz (1983a and 1983b), Twincham et al. (1984), Nelson and Glatz (1985), Brown and Glatz (1987), and Chen and Rohani (1992). Grabenbauer and Glatz (1981), Glatz et al. (1986), Rohani and Chen (1993), and Raphael and Rohani (1996a) have used mixed suspension mixed product removal precipitators (MSMPRP). Only three studies have been reported on tubular precipitators: Virkar et al. (1982), Chan et al. (1986), and Raphael et al. (1995).

Previous studies on protein precipitation (Virkar et al., 1982; Grabenbauer and Glatz, 1981; Petenate and Glatz, 1983a and 1983b; Glatz et al., 1986), including the present study, have shown that, protein particles can exhibit a unimodal or bimodal PSD in population density depending on the type of the precipitator and the operating conditions used. The presence of large aggregates together with smaller aggregates and primary particles may result in a bimodal PSD. In such a system, determination of kinetic parameters is difficult, because the number of particles changes fast, due to aggregation and large particles cannot be distinguished from aggregates.

Virkar et al. (1982) studied the isolectric precipitation of soy protein in a tubular precipitator under turbulent flow regime. The flow Reynolds number used was greater than 20000, with the flow of protein solution being much greater than that of the acid. The authors modeled the growth of protein particles along the tubular precipitator using the Smoluchowski’s theory. They assumed that the particle growth is by orthokinetic aggregation of particles caused by particle particle collision in the turbulent flow regime. That is, for two particles of radius \( r_i \) and \( r_j \) the frequency of orthokinetic collision per unit volume in turbulent shear flow is given by:

\[
F_{ij} = \frac{4}{3} \left( r_i + r_j \right)^3 \left( \frac{N_i N_j}{v} \right) \left( \frac{\eta}{v} \right)^{1/2} \left( 1 - \frac{\delta_{ij}}{2} \right) \quad (1a)
\]

where the energy dissipation per unit mass was defined as:

\[
\frac{\eta}{v} = 2 \rho_o u_l d_p \quad (1b)
\]

For effective collision leading to a lasting aggregation, the radius of a newly formed aggregate was assumed to be less than \( r_{max} (30 \mu m) \) otherwise the collision was ineffective. That is:

\[
\left( r_i^3 + r_j^3 \right)^{1/3} > r_{max} \quad \Lambda_{ij} = 0 \quad \text{ineffective collision} \quad (1c)
\]

\[
\left( r_i^3 + r_j^3 \right)^{1/3} \leq r_{max} \quad \Lambda_{ij} = 1 \quad \text{effective collision} \quad (1c)
\]
The model results showed good predictions between the calculated and experimental mean particle sizes for the initial stages (at the start). As the mean residence time (growth time) increased the calculated mean size was much larger than the experimental data. This was attributed to the lack of a proper model for the breakage of the larger aggregates (formed by the collisions that resulted in particles larger than \( r_{max} \)), and due to the exclusion of the precipitation kinetics: nucleation and particle growth rates.

Virkar and co-authors considered the tubular precipitator as an aggregator with a limited particle breakage. Also, the assumption that the precipitation process is a very fast reaction and is almost complete within 1 s, neglects the molecular particle growth by diffusion which may continue beyond 1 s. This is because the diffusion process is a slow process and depends on the mixing of the suspension. Thus, the particle growth may continue for some time before the concentration gradient vanishes, although the contribution in the particle size change by the growth mechanism may be much smaller when compared to that caused by the aggregation mechanism.

Chen et al. (1986) studied the kinetics of soy protein in a tubular precipitator using different precipitants. Experiments were carried out in the laminar flow regime \((Re = 500, \text{ based on total feed flow rate})\). In their study, no attempt was made to determine the precipitation kinetics, or to model the PSD for the tubular precipitator. They also reported that, for high concentrations of protein solutions, the solid protein comes out of the solution within 1 s. The initial primary particles grow by aggregation and their final size depends on shear forces and the type of the precipitant used.

In another study, Rivera and Randolph (1978) studied the continuous precipitation of pentaerythritol tetranitrate (TNT) in a short tubular precipitator. The precipitator was assumed to operate in the dispersed plug flow regime \((Pe = 51)\) with no particle aggregation or breakage. The PBE was given as:

\[
D_e \frac{d^2 \rho}{d z^2} = u_e \frac{d \rho}{d z} + G \frac{d \rho}{d L} \quad \text{.................................. (2)}
\]

The PBE was solved using the method of moments. The nucleation rate \((B^0)\) and the growth rate \((G)\) along the tubular precipitator were assumed to be functions of the initial values: \(B^0 = B^0(1-z)^\sigma\) and \(G = G^\sigma(1-z)^\gamma\). A similar method was used by Raphael et al. (1995), for determination of the precipitation kinetic parameters for the sunflower protein from the tubular precipitator (using the data in the dispersion region). In their study, the deconvolution of moments to generate the PSD using the matrix inversion method suggested by Randolph and Larson (1988) was not successful. Instead, a method suggested by Hulburt and Katz (1964) using gamma weighted Laguerre polynomials was used.

In the present study, a full model incorporating primary nucleation, molecular particle growth, particle aggregation, particle breakage, and flow dispersion along the tubular precipitator is being used. The experimental data from the tubular and batch precipitators at known conditions were used to obtain the kinetic parameters by optimization technique. Given the operating parameters: protein feed concentration, feed flow rate, dispersion rate, and particle size distribution at known locations after the mixing zone, predictions of the PSD and solids yield of sunflower protein (during precipitation at isoelectric pH using aqueous HCl solution) along the tubular precipitator can be calculated.

### Methods of solving the population balance equation (PBE)

In order to determine the precipitation kinetics (from experimental data), it is necessary to represent the system using the population balance equation (PBE). The population balance equations for crystallization and precipitation processes in the batch precipitator and mixed-stirred precipitators have been solved by several workers. The solution methods can be categorized into six main groups: analytical method, method of moments, method of weighted residue (MWR), method of classes, stochastic method, and finite differences (discretization) method. Table 1 presents a summary of the these methods.

For a simple linear PBE, Laplace transformation or inverse transformation has been used to give an approximate solution. Saleeby and Lee (1994) solved analytically the PBE for an MSMPR crystallizer with agglomeration. When nonlinear terms (aggregation or breakage) are included in the PBE, solution by analytical method becomes difficult.

Hulburt and Katz (1964) formulated the number density balance equation for continuous flow systems using a classical statistical mechanics method. They suggested an analytical method of moments for solving the balance equation, then gamma weighted Laguerre polynomials were used to reconstruct the PSD from the moments. This deconvolution method has been reported to result in strong oscillations of the PBE (Randolph and Larson, 1988). Instead, Randolph and Larson (1988) proposed the matrix inversion method. In this method, the moments equations are transformed into linear algebraic equations and solved numerically.

The method of moments developed by Randolph and Larson (1988), is widely used to solve PBE for MSMPR precipitators. The method transforms the difficult to solve integro-differential equation into a set of ordinary differential equations by

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**Table 1**

<table>
<thead>
<tr>
<th>Solution Method</th>
<th>Reference</th>
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<tbody>
<tr>
<td>Inverse transformation</td>
<td>Saleeby and Lee (1994)</td>
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<td>MWR with shifted Legendre functions</td>
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<td>MWR</td>
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<td>Palfranks and orthogonal collocation</td>
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<tr>
<td>Orthogonal collocation</td>
<td>Lakatos et al. (1984)</td>
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<td>MWR with fractional moments</td>
<td>Bhattacharyya and Chakraborty (1992)</td>
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<td>Method of classes</td>
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<td>Stochastic method</td>
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<td>Finite differences (discretization) method</td>
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<td>Kimand Tarbell (1991)</td>
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moment method. The drawback of this method lies in the deconvolution of the moments to yield the PSD, sometimes convergence to the final solution is not achieved, resulting in positive and negative number population densities (Kim and Tarbell, 1991 and Raphael et al., 1995).

The method of weighted residue (MWR), after Finlayson (1972), has been used to solve the PBE by several workers. The method involves expanding the unknown solution using a combination of known trial functions (Laguerre Polynomial, shifted Legendre polynomials, or problem specific polynomials) and substituting into the original equation. The satisfying solution is obtained when the residual is zero, or close to zero. This is achieved by orthogonalizing the residual with a set of weighting functions. The difficult part of this method is in identifying the form of the trial function which represents the solution in the whole range. Orthogonal Collocation is a special form of MWR, in that the resulting residual is set to zero at fixed node points.

Padia and Bhatia (1991) analyzed the stability of the precipitation systems resulting from the coupling of the material balance equation, PBE, and kinetic equations for nucleation and growth rates. They stated that the solution of these coupled equation exhibit the multiple steady states when certain ranges of kinetic parameters are used. Similar observations have been reported by Witkowski and Rawlings (1987) and Tavare and Garside (1985).

In the current study, the orthogonal collocation method was used to solve the PBE in the particle size direction and multiple shooting method in the flow direction. The exponentially weighted Laguerre polynomial was used as a trial solution.

**Experimental method**

In this study, experiments on isoelectric precipitation of sunflower protein were carried out in a tubular precipitator. The effects of flow velocity, mean residence time, and protein feed concentrations on PSD were studied. The tubular precipitator consisted of a horizontal 20 m long, 6 mm inner diameter glass tube, having four 20 \( \times \) 11 mm Kenics static mixers (Koch Engineering Co. Inc., Wichita, KA) at the entrance and 4 sampling ports located at 0.4, 3.3, 9.5, and 20.1 m from the entrance (Figure 1). The flow rates in the tubular precipitator were set to correspond to flow \( Re = 800 \) or \( Re = 5000 \), based on the total feed flow rates (see Table 2).

The axial dispersion number in the tubular precipitator 
\[ Re = \frac{u_D}{D} \]
was determined in terms of Peclet number (\( Pe = u_D/D \)) using the residence time distribution (RTD) method. Pulse response from a non-reactive tracer (aqueous KCl) was used for this study at different flow velocities, Levenspiel (1972). The resulting best fit equation was given by:

\[ \frac{1}{Pe} = 0.0263 - 0.00015u \_D \]  \hspace{1cm} (3a)

where \( u_D (\text{cm/s}) \) is the superficial liquid flow velocity and the coefficient of regression, \( R^2 = 0.905 \). The range of the flow velocity covered both laminar and turbulent flow regimes. The dispersion number was assumed to remain constant along the length of the tubular precipitator at a given flow velocity. According to Nauman and Burcham (1983), the axial dispersion model can be used to approximate the laminar flow systems, even though the flow velocity profiles are not flat. In the present work, the dispersion coefficient \( D \) combines the effect of laminar velocity profiles and the molecular diffusion (Taylor-Aris dispersion). This is valid for long tubes with small ratio of \( D/u\_D \).

For all experiments, the protein feed solution was extracted from defatted seeds (prepared from confectionery-type sunflower seeds) using aqueous solution (1:20 g/mL) at pH 10. Precipitation was carried out at 22 ± 2°C at pH 4 (isoelectric pH) by contacting equal volumes of protein solution with aqueous HCl solution. The concentration of protein in the feed solution was maintained at 2.8 or 11.8 kg/m³, while the concentration of aqueous HCl solution was predetermined by titration (in order to achieve a final solution pH of 4). A total of four experiments with replications were performed. Two more runs were carried out, one at an intermediate flow Reynolds number (\( Re = 3000 \)) and the other at an intermediate protein concentration (7.1 kg/m³).

In a separate study, batch precipitation experiments were performed to determine the final concentration of protein in the solution (\( C \)) after precipitation (using equal volumes of protein solution and aqueous HCl at temperature range between 295 and 318 K). The final protein concentration was found to be related to the precipitation temperature (\( T_p \), in K) and the initial concentration of protein (\( C_0 \)) by:

\[ C^+ = K_s \frac{C_0}{2} \]  \hspace{1cm} (3b)

\[ K_s = k_0 e^{k_{PE}T_p} \]  \hspace{1cm} (3c)

Equations (3b) and (3c) were fitted to the experimental data using least squares error method (SAS manual, 1985). The optimized parameters obtained were: \( k_0 = 0.785 \) and \( k_{PE} = 247.6 \) K with \( R^2 = 0.992 \).
Figure 2 — Effect of flow Reynolds number on the mean particle size: feed ratio = 1.0 v/v (aqueous HCl; protein solution) and protein feed concentration = 11.8 kg/m³.

Figure 3 — Effect of protein feed concentration on the mean particle size: A) total flow Re = 5000, B) total flow Re = 800; feed ratio = 1.0 v/v (aqueous HCl; protein solution), and protein feed concentration = 11.8 kg/m³.

The PSD of the precipitate was obtained using a Coulter counter (model TAI1, Coulter Electronics, Inc., Hialeah, FL). The solids protein concentration was measured using an on-line turbidimeter, Raphael and Rohani (1996b). Details of the experimental method and measurements of the dispersion along the tubular precipitator can be found in Raphael et al. (1995).

Experimental results

Nucleation takes place in the mixing zone (entrance zone with the static mixers) where the supersaturation is high. Beyond the entrance zone, the supersaturation decreases sharply. The nuclei continue to grow by diffusion to form the primary protein particles. The primary particles aggregate to form larger particles. The maximum stable size of the aggregates is governed by the shear forces exerted by the surrounding fluid. Since the protein supersaturation drops sharply in the tubular precipitator, molecular growth by protein deposition does not contribute significantly to the overall growth (Glatz et al., 1986; Virkar et al., 1982).

The tubular precipitator operating in the laminar flow regime resulted in larger mean particle sizes, which increased with increases in mean residence time (Figure 2). This is due to the continuous growth of smaller particles,
mainly by an aggregation process. Low fluid-induced shear rate in the laminar flow regime favors the formation of the large aggregates. In the turbulent flow regime, the mean particle sizes were small and varied little with mean residence time. The effect of protein feed concentration on the mean particle sizes (for Re = 800 and 5000) is presented in Figure 3. Increasing the protein feed concentration under turbulent flow regime (Figure 3A), increased the solids concentration. This resulted in increased particle-particle collisions leading to slightly smaller mean particle sizes due to breakage. In the laminar flow regime (Figure 3B) particle-particle collisions were not high enough to cause significant breakage, and an overall increase in the particle mean size was observed. Figure 4 shows the measured population densities of the precipitate at four sampling ports along the tubular precipitator. For the turbulent flow regime, the population density distribution showed a uni-modal distribution with a narrow spread (Figure 4A). The spread of the PSD decreased with an increase in the mean residence time. This suggests that, the rate of aggregate breakage was higher than the rate of aggregate formation. Close to the precipitator entrance, the dispersion number approaches that of a mixed vessel. As the flow becomes fully developed (ca. 10d), the precipitator approaches the plug flow behavior. In the laminar flow regime, the PSD showed a wide spread (Figure 4B), resulting in a high coefficient of variation (CV) defined in Equation (25). This is attributed to the presence of backmixing, low particle-particle collision energy, and low fluid-induced shear which are favorable conditions for aggregation. In summary, fluid-induced shear and the flow dispersion determine the steady-state PSD.

The amount of recovered solid protein in both laminar and turbulent regimes increased with an increase in the mean residence time (Figure 5). This shows that the solid protein would come out of the solution by molecular diffusion along the length of the precipitator even at low super-saturation levels. Also, increasing the feed concentration resulted in an increase in solids concentration due to a high degree of supersaturation.

**Modeling of a tubular precipitator with particle aggregation and breakage**

At steady state, the population balance equation for a tubular precipitator in the presence of aggregation, breakage, and secondary nucleation (Randolph and Larson, 1988) is given by:

\[
\begin{align*}
D_z \frac{\partial^2 p}{\partial z^2} - u_z \frac{\partial p}{\partial z} - \frac{\partial}{\partial z}\left(G \frac{\partial p}{\partial L}\right) + (B_a - D_a) + (B_b - D_b) \\
+ \beta_0 \delta(L - L_o) = 0
\end{align*}
\]

where \(p\) (the population density) is a function of particle size \(L\) and position \(z\) along the tubular precipitator. \(D_z\) is the axial dispersion coefficient, \(u_z\) is mean flow velocity, \(G\) is the molecular particle growth rate, \((B_a - D_a)\) is the net rate of particle aggregation, \((B_b - D_b)\) is the net rate of particle breakage, and \(\beta_0\) is the secondary nucleation rate of particles of size \(L_o\). Equation (4) assumes no radial dispersion and has initial condition; \(p(0, z) = J/G\) as \(L \to 0\) (\(J\) is the primary nucleation rate which is assumed to be the dominant nucleation mechanism). The boundary condition at the entrance is given by: \(u_z p(L, 0) + D_z \frac{dp(L, 0)}{dz} = 0\) that is, no seeding or solids at the entrance. At the exit of the precipitator, the boundary condition is \(dp/dz = 0\). In this study the radial dispersion was neglected because the ratio of the tube diameter to the length was very small (3 \(\times\) 10\(^{-4}\)).

**Dimensionless Parameters**

Introducing the dimensionless parameters defined as:

\[
\zeta = \frac{z}{z_T}, \quad x = \frac{L}{L_o}, \quad \text{and} \quad \gamma(x, \zeta) = p(x, \zeta)N_o = \frac{L}{L_o} p(L, \zeta)\frac{N_o}{N_o}
\]

where \(Pe(\zeta)\) is the dimensionless distance along the tubular precipitator, \(z_T\) is the total length of the tubular precipitator, \(x\) is dimensionless particle size, and \(L_o\) is the maximum measurable particle size divided by the largest root of the Laguerre polynomial. \(L_o = 3.528 \mu m\) is used to scale down the experimental particle size range to be within the range of the zeros of the Laguerre polynomial. \(\gamma(x, \zeta)\) is the dimensionless population density. \(N_o\) is a constant value (number density) used to reduce the magnitude of population density. In this study \(N_o\) was set at 10\(^5\). Substituting dimensionless parameters into Equation (4), and assuming negligible secondary nucleation and that particle growth rate is independent of particle size, Equation (4) becomes:

\[
\frac{\partial \gamma}{\partial x} - \frac{Pe(\zeta) \partial \gamma}{\partial \zeta} - \frac{Pe(\zeta) \gamma}{L_o} \frac{G \partial \gamma}{L_o} + \frac{Pe(\zeta) \gamma}{L_o} \frac{L}{L_0} = 0
\]

where \(Pe(\zeta)\) is the Peclet Number \((u_z z_T/D_z)\). As \(Pe(\zeta) \to 0\) the flow approaches the mixed flow regime and as \(Pe(\zeta) \to \infty\) the flow approaches the plug flow regime. In dimensionless variables the initial and boundary conditions are:

\[
\gamma(x, \zeta) \to 0, 0 \leq \zeta \leq 1 \quad \text{(7a)}
\]

\[
\gamma(x, 0) - \frac{1}{Pe(\zeta)} \frac{dy(x, 0)}{d\zeta} = 0, \quad \zeta = 0, x > 0 \quad \text{(7b)}
\]
\[ \frac{d\phi(x, \zeta)}{d\zeta} = 0; \quad \zeta = 1, x > 0 \quad \ldots \quad (7c) \]

**Primary Nucleation Rate (J)**

The formation of new particles is assumed to be a continuous process which occurs throughout the precipitator. However, the rate of nucleation decreases as the fluid moves downstream. The nucleation rate along the tubular precipitator is expressed as a function of the initial nucleation rate \( j_0 \), that is, the nucleation rate at the precipitator entrance. At the entrance, there is a high degree of supersaturation leading to high nucleation rate. As the degree of supersaturation decreases along the tubular precipitator, so does the nucleation rate. The nucleation rate is expressed as:

\[ J = J_0 \left(1 - k_g \zeta^g\right) \quad \ldots \quad (8a) \]

The parameter \( k_g \) determines the rate of depletion in the supersaturation along the precipitator. The initial nucleation rate is determined using the homogeneous nucleation theory for fast reactions (Mullin, 1972) as:

\[ J_0 = k_{Bo} \exp \left(-\frac{k_m}{(\ln S_0)^2}\zeta^g\right) \quad \ldots \quad (8b) \]

where the supersaturation ratio is \( S_0 = C_0/C^* \), and \( k_m \) and \( k_{Bo} \) are constants. \( C_0 \) and \( C^* \) (calculated from Equations 3a and 3c) are the protein feed concentration and the equilibrium protein concentration, respectively. Equation (8b) predicts a rapid increase in the nucleation rate with increases in the degree of supersaturation \( S_0 \).

Then Equation (8a) becomes:

\[ \dot{J}(\zeta) = k_{Bo} \exp \left(-\frac{k_m}{(\ln S_0)^2}\right)(1 - k_g \zeta^g) \quad \ldots \quad (8c) \]

**Growth Rate (G)**

Molecular growth rate of particles due to deposition of protein from the solution on the particle surface is modeled using a similar argument as for the nucleation rate. The growth rate by molecular mechanism increases with an increase in the concentration gradient. Near the entrance, the growth rate is high and decreases as the concentration gradient decreases along the tubular precipitator. Therefore, the growth rate is expressed as a function of the initial growth rate \( G_{o^*} \):

\[ G(\zeta) = G_{o^*} (1 - k_g \zeta^g) \quad \ldots \quad (9a) \]

where \( G_{o^*} \) is defined as:

\[ G_{o^*} = k_{Go} \sigma_{o^*}^g \quad \ldots \quad (9b) \]

The relative supersaturation at \( \zeta = 0 \) is given by: \( \sigma_{o^*}^g = (C_0 - C^*)/C^* \). The parameter \( g \) determines the mechanism governing particle growth. When \( g = 1 \), the particle growth is diffusion controlled, \( 1 < g < 2 \) implies that screw dislocation mechanism controls the particle growth, and when \( g > 2 \) polynuclear mechanism controls the particle growth. Substituting Equation (9b) into Equation (9a) yields:

\[ G(\zeta) = k_{Go} \sigma_{o^*}^g (1 - k_g \zeta^g) \quad \ldots \quad (9c) \]

**Particle Breakage Rate**

The particle breakage rate term, \( B = (B_h - D_h) \), is assumed to be a two equal body volume-conserved breakage (Glatz et al., 1986). That is, a breaking particle (death) results in the birth of two particles of equal volume. This simple model is given as:

\[ D_h(L) = k_d L^d p(L, \zeta) \quad \ldots \quad (10) \]

\[ B_h = 2 D_h(2^{1/3}L, \zeta) \quad \ldots \quad (11) \]

Then the net rate of particle breakage, \( B \), is given as:

\[ B = k_d L^{d+1} p(1.26L, \zeta) - p(L, \zeta) \quad \ldots \quad (12) \]

Introducing the dimensionless variables into Equation (12), results in:

\[ B = k_d \frac{N_o}{L_c} (x L_c)^d \left[2^{d+1} y(1.26x, \zeta) - y(x, \zeta)\right] \quad \ldots \quad (13) \]

**Particle Aggregation Rate**

The particle aggregation rate is modeled using the formulation of Hulburt and Katz (1964). The death and birth of particles due to aggregation is given by:

\[ A(L) = (B_d - D_d) = \int_0^L \left[ \beta(L_1, L_2) \frac{p(L_1, \zeta)}{L_2^2} dL_1 \right] \left[1 - \frac{p(L_1, \zeta)}{p(L, \zeta)} dL_1 \right] \quad \ldots \quad (14) \]

where \( L_1^3 = L_2^3 \). Because of the limited knowledge of aggregation kernel, \( \beta(L_1, L_2) \), it is assumed to be constant, \( \beta_o \), and independent of the particle size. Introducing the dimensionless variables into Equation (14) yields:

\[ A(\zeta) = P_A \frac{\chi^2}{2} \left[\int_0^{\chi_0^3} \left(y(\zeta) - y(\zeta)\right) dy(\zeta)\right] \quad \ldots \quad (15) \]

where \( P_A \) is the polynuclear mechanism controlling the particle growth. The collocation method allows the use of uneven mesh in the direction while the MSM is applied on the even grid.
mesh. The procedure is to assume a Lagrange polynomial as a solution of the PBE. This trial solution is selected from a set of functions defined in the domain \([0, \infty)\). For this reason, the Laguerre functions were selected. The trial solution is defined as:

\[
y_n(x, \zeta) = \sum_{i=1}^{n} \alpha_i(\zeta) \Phi_{i-1}(x) \quad \cdots \quad (16)
\]

where \(\alpha(\zeta)\) is a function of \(\zeta\) and \(\Phi(x)\) is a function of \(x\). In Equation 16, \(y_n\) is a \((n-1)\)th polynomial in \(x\).

The function \(\Phi(x)\) is defined as:

\[
\Phi_l(x) = e^{-x} \psi_l(x) \quad \cdots \quad (17)
\]

\(\psi_l(x)\) are the Laguerre polynomials. The \(n\)th Laguerre polynomial (Courant and Hilbert, 1966) is defined as:

\[
\psi_n(x) = \sum_{j=0}^{n} \frac{(-1)^j (n!)^2 x^j}{\beta (n-j)!} \quad \cdots \quad (18)
\]

with \(\psi_0(x) = 1\), \(\psi_1(x) = -x + 1\), and \(\psi_2(x) = x^2 - 4x + 2\). These polynomials are orthogonal with respect to the exponential weighting in the range \([0, \infty)\). The solution procedure is to substitute Equation (16) in the PBE. Since the expansion solution is an approximation, a non-zero residual, \(R(x, \zeta)\) will result after this substitution. The basis of the orthogonal collocation method is to set this residual to zero at all collocation points, \(x_i\). This is done by determining the coefficients \(\alpha_i(\zeta) (i = 0, 1, 2, ..., n-1)\) so that \(R(x_i, \zeta) = 0\). For an "exact" solution when using the orthogonal collocation method, it is required that the collocation points \(x_i\) be the roots (zeros) of the \(n\)th order Laguerre polynomial (Villadsen and Michelsen, 1978). Substituting the approximate solution into the PBE and introducing the matrix notation (see Appendix) results in:

\[
\frac{d^2}{d\zeta^2} - \text{Pe}(\zeta) \frac{d\alpha}{d\zeta} = \text{Pe}(\zeta) \frac{G_{\text{m}}}{L_0} \left( \Phi \frac{d}{d\zeta} \right) - \frac{e^{-\zeta} \Phi}{\zeta} + \frac{\zeta}{\text{Pe}(\zeta)} \frac{d}{d\zeta} \left[ \bar{R}(x, \zeta) \right] \quad \cdots \quad (19)
\]

where \(\bar{R}(x, \zeta)\) is the residue matrix. Therefore, the PBE has been transformed into a set of second order ordinary differential equations (boundary value problem). The corresponding boundary conditions in matrix form are:

\[
\begin{align}
\Phi \alpha(\zeta) & \bigg|_{x=0} = \frac{J / N_0}{G / L_0} ; \quad x \rightarrow 0, 0 \leq \zeta \leq 1 \quad \cdots \quad (20a) \\
P e(\zeta) \frac{\alpha(\zeta)}{\zeta} - \frac{d\alpha}{d\zeta} & = 0 ; \quad \zeta = 0, x > 0 \quad \cdots \quad (20b) \\
\frac{d\alpha}{d\zeta} & = 0 ; \quad \zeta = 1, x > 0 \quad \cdots \quad (20c)
\end{align}
\]

**SIMULATION OF THE MODEL AND SENSITIVITY ANALYSIS**

In order to determine the structural accuracy of the overall model and its behavior to changes in the kinetic parameters, it was necessary to carry out a detailed sensitivity analysis. Due to the lack of experimental data in the literature, the PBE was solved using a hypothetical PSD derived from skewed normal distribution curves. This sensitivity analysis would also demonstrate the flexibility and the convergence of the proposed numerical scheme using different model parameters on the PSD profiles. The optimum model parameters which satisfy the boundary conditions (BC) were determined by the optimization method. The dimensionless PSDs at the port near the entrance, \(\zeta = 0.02\), and at some other point, \(\zeta = 0.5\), along the precipitator length, were assumed to be continuous and represented by simple distribution equations. At the entrance the PSD was defined as:

\[
y(x, \zeta_0) = \frac{K_0}{\mu_o} \exp\left[-\left(\frac{x-x_0}{\mu_o}\right)^2\right] \quad \cdots \quad (21)
\]

where \(x_0\) is assumed to be the dimensionless nuclei size (very small number = 0.001), \(\mu_o = 1\), \(K_0 = 10\) and \(\zeta_0 = 0.5\). The density function distribution was assumed as:

\[
y(x, \zeta_e) = \frac{K_e}{\mu_e} \exp\left[-\left(\frac{x-x_e}{\mu_e}\right)^2\right] \quad \cdots \quad (22)
\]

**Figure 6** — Reference PSD and simulated PSD profiles obtained by varying the breakage rate parameter \(d\).
TABLE 3
Values of Parameters Used in the Model Simulation
(Reference Data)

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nucleation rate:</td>
<td>( n = 3.5 )</td>
</tr>
<tr>
<td></td>
<td>( k_1 = 0.5 )</td>
</tr>
<tr>
<td></td>
<td>( k_2 = 0.30 )</td>
</tr>
<tr>
<td></td>
<td>( k_3 = 5 \times 10^{-6} )</td>
</tr>
<tr>
<td>Growth rate:</td>
<td>( b = 2.0 )</td>
</tr>
<tr>
<td></td>
<td>( k_{Bo} = 1 \times 10^6 \text{ no/mL-s} )</td>
</tr>
<tr>
<td></td>
<td>( k_{Go} = 4 \times 10^{-5} \text{ µm/s} )</td>
</tr>
<tr>
<td>Breakage rate:</td>
<td>( d = 0.2 )</td>
</tr>
<tr>
<td>Aggregation rate:</td>
<td>( \beta_0 = 1.0 \times 10^{-15} \text{ µm}^4 \text{ mL}^{-1} \text{ s}^{-1} )</td>
</tr>
<tr>
<td>Protein conc.:</td>
<td>( C_o = 12.0 \text{ kg/m}^3 )</td>
</tr>
<tr>
<td>Flow velocity:</td>
<td>( u_z = 0.13 \text{ m/s} )</td>
</tr>
<tr>
<td></td>
<td>( C^* = 2.0 \text{ kg/m}^3 )</td>
</tr>
</tbody>
</table>

Figure 7 — Simulated PSD profiles obtained by varying the nucleation rate parameter \( k_B \).

where \( x \) was assumed to be the dimensionless exit mean particle size \((= 11)\), \( \mu_x = 1.41 \times x_c CV \), \( K_x = 10 \pi^{1/2} \), and \( CV \) was 0.45. The two PSD are shown in Figure 6A. The entrance (port 1) PSD shows the existence of many small particles due to homogeneous nucleation. Then, it is assumed that after sometime, the slurry will have moved a distance \( c = 0.5 \) along the tubular precipitator. During this time, nucleation, particle growth, particle aggregation, and particle breakage take place. Using the two PSDs as boundary conditions, along with the assumed values of parameters given in Table 3 (not optimized), the PBE was solved using OCMSM to generate the intermediate PSD profiles (no intermediate constraints). When the breakage rate is increased, by increasing the constant \( k_B \) or the parameter \( d \), the PSD shifts towards the small particle size range (see Figure 6B), indicating that the population density of fine particles has increased. Decreasing the number of initial nuclei by increasing parameter \( a \) or \( k_B \), leads to a fast depletion of the supersaturation. This results in a low population of the small particles along the tubular precipitator, due to the absence of the nucleation process (Figure 7A). Whereas, if the value of \( k_B \) is made zero (slow depletion of supersaturation), the nucleation process becomes very slow and takes place throughout the tubular length. This, together with particle breakage results in high values of fine particles (Figure 7B).

Figure 8A shows the PSD profiles generated by varying the growth rate exponent. Growth rate is associated with the molecular growth of the small particles and primary particles.
before they join with larger aggregates. Increasing the growth rate exponent \( g \) results in multi-modal PSD. This shows the presence of mixed sizes of particles, the aggregates and large particles formed by molecular growth. Larger values of the growth exponent \( g \) leads to multiple solutions of the PBE solution.

Increasing the flow velocity (high Reynolds number) shows that the PSD does not form a multi-modal distribution (Figure 8B). This might be due to high shear rate caused by turbulent flow resulting in breakage of the intermediate sized aggregates or that, the particle-particle collisions are not effective in forming a lasting aggregate.

These parameter sensitivity studies, as illustrated in these examples, shows that the numerical scheme is flexible enough to generate the PSD profiles to match the experimentally obtained PSD (boundary conditions). A similar scheme was used during parameter optimization where four different data sets (each with 2 inner constraints) were used to determine the best set of parameters resulting in the minimum residue.

**Determination of Precipitation Kinetic Parameters from Experimental Data**

A total of 11 parameters from model equations (nucleation rate equation parameters: \( a, k_B, k_{Bo}, \) and \( k_B \); growth rate equation parameters: \( b, k_G, k_{Go}, \) and \( g \); breakage rate equation parameters: \( d, \) and \( k_d \); and aggregation rate equation parameter: \( \beta_a \)) were determined by the optimization method. The estimated parameters were then substituted into the respective equation for solving the population density equation along the tubular precipitator. The calculated population density data were used to determine the local mean particle size and the coefficient of variation (corresponding to experimental sampling ports). Parameter optimization was done using the Simulated Annealing Method (SAM), Goffe et al. (1994), for determination of the global parameters. The method is slow, but sure. The Multiple Shooting Method (MSM) was used to obtain the PSD profiles between the two boundary conditions, the entrance port and the exit port, using the iterative method. During each calculation pass, the boundary conditions were satisfied. The PSD for the interior ports (2 and 3) were represented using the mean particle size and the coefficient of variation. These interior data were used as constraints for the optimization equation, that is, the optimum parameters should satisfy the two boundary conditions profiles and the interior distributions at ports 2 and 3. Four sets of experimental data were used for this optimization (Table 4). The objective function to be optimized was given as:

\[
\min H = \sum_{i=1}^{3} \left[ \frac{(dm_{ji} - 1)}{d_{ji}} \right]^2 + \left[ \frac{(CVm_{ji} - 1)}{CV_{ji}} \right]^2 \quad \ldots \quad (23)
\]

where \( P \) is a matrix of parameters, \( [a k_B k_{Bo} b k_G k_{Go} g d k_d [\beta_a]] \), \( dm_{ji} \) and \( CV_{ji} \) are the calculated values of the mean particle size and the coefficient of variation for run \( i \) at port \( j \), respectively. \( d_{ji} \) and \( CV_{ji} \) are the corresponding experimental values. \( dm_{ji} \) and \( CV_{ji} \) were calculated as:

\[
\frac{\sum_{j=1}^{n} L_j^d p_j \Delta L_j}{\sum_{j=1}^{n} L_j^d p_j \Delta L_j} \quad \ldots \quad (24)
\]

### Tables

**Table 4** Summary of the Runs in the Tubular Precipitator used in the Determination of Kinetic Parameters, Experimental Conditions and Measured Mean Particle Size and CV of Samples from Ports 2 and 3

<table>
<thead>
<tr>
<th>Run No.</th>
<th>( C_{in} ) (kg/m³)</th>
<th>( C^* ) (kg/m³)</th>
<th>( u_i ) (m/s)</th>
<th>( d_{m1} ) (μm)</th>
<th>( d_{m2} ) (μm)</th>
<th>( CV_{m1} ) (%)</th>
<th>( CV_{m2} ) (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>L1</td>
<td>11.8</td>
<td>1.9</td>
<td>0.13 (800)</td>
<td>27.8 ± 1.1</td>
<td>30.7 ± 1.7</td>
<td>38.5</td>
<td>43.1</td>
</tr>
<tr>
<td>L2</td>
<td>11.8</td>
<td>1.9</td>
<td>0.83 (5000)</td>
<td>8.6 ± 1.1</td>
<td>7.9 ± 1.7</td>
<td>24.3</td>
<td>23.1</td>
</tr>
<tr>
<td>L6</td>
<td>2.8</td>
<td>0.47</td>
<td>0.13 (800)</td>
<td>22.8 ± 1.1</td>
<td>27.6 ± 1.7</td>
<td>35.5</td>
<td>28.6</td>
</tr>
<tr>
<td>L7</td>
<td>2.8</td>
<td>0.47</td>
<td>0.83 (5000)</td>
<td>10.2 ± 1.1</td>
<td>9.5 ± 1.7</td>
<td>22.1</td>
<td>21.8</td>
</tr>
</tbody>
</table>

**Table 5** Initial Estimates of Kinetic Parameters, the Search Range Used in Optimization, and the Final Optimum Values

<table>
<thead>
<tr>
<th>Equation</th>
<th>Parameter</th>
<th>Min. value</th>
<th>Max. value</th>
<th>Initial value</th>
<th>Optimum value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nucleation rate</td>
<td>( a )</td>
<td>0.5</td>
<td>10.0</td>
<td>1.0</td>
<td>2567</td>
</tr>
<tr>
<td></td>
<td>( k_B )</td>
<td>0.001</td>
<td>0.98</td>
<td>0.5</td>
<td>353</td>
</tr>
<tr>
<td></td>
<td>( k_{Bo} )</td>
<td>1.0</td>
<td>10¹⁴</td>
<td>10⁶</td>
<td>3850 × 10¹⁰</td>
</tr>
<tr>
<td>Growth rate</td>
<td>( b )</td>
<td>0.0</td>
<td>5.0</td>
<td>1.0</td>
<td>1842</td>
</tr>
<tr>
<td></td>
<td>( k_G )</td>
<td>0.001</td>
<td>0.98</td>
<td>0.5</td>
<td>1999</td>
</tr>
<tr>
<td></td>
<td>( k_{Go} )</td>
<td>0.0</td>
<td>1.0</td>
<td>0.1</td>
<td>5.107 × 10⁴</td>
</tr>
<tr>
<td></td>
<td>( g )</td>
<td>0.0</td>
<td>5.0</td>
<td>1.0</td>
<td>1512</td>
</tr>
<tr>
<td>Breakage rate</td>
<td>( d )</td>
<td>0.1</td>
<td>5.0</td>
<td>1.0</td>
<td>784</td>
</tr>
<tr>
<td></td>
<td>( k_d )</td>
<td>0.0</td>
<td>5.0</td>
<td>0.1</td>
<td>2744 × 10⁶</td>
</tr>
<tr>
<td>Aggregation rate</td>
<td>( \beta_a )</td>
<td>0.0</td>
<td>10⁻⁹</td>
<td>10⁻¹²</td>
<td>9549 × 10⁻¹⁶</td>
</tr>
</tbody>
</table>

Final value of \( H = 0.577 \)
Figure 9 — A) Model fit to the experimental data using the optimum parameters; B) Model prediction of the experimental data.

Table 5 shows the list of the initial estimates and the search range used for optimization. Some of these data were based on the knowledge from previous studies (Raphael et al., 1995 and Raphael and Rohani, 1996a).

The numerical algorithm starts by entering the scaled down PSD for the boundary conditions at selected node points (the roots of the trial function). This was achieved by fitting the experimental PSD data using a cubic spline polynomial, then interpolated at the desired nodes. The interpolated data were checked for accuracy by recalculating the mean particle size and the CV, then compared with the original data. In all interpolated data used in this parameter estimation, the accuracy was within ±8% for the mean particle sizes smaller than 10 μm, and within ±5% for the mean particle sizes larger than 10 μm. Interpolated CV data showed discrepancies of up to ±50% because the roots of the Laguerre polynomial did not match the experimental measuring sizes (many roots were in the lower end and sparsely distributed in the higher end). Then, the initial guess of parameters for the optimization routine were entered. The optimization program (Simulated Annealing Method) solved the objective function value by calling the BVP-ODE solver with 101-equally spaced node points in ζ direction (along the tubular precipitator). The BVP-ODE solver (Non-linear Multiple Shooting Method (MSM) by Ascher et al., 1988) called the other subroutines which calculated the net aggregation and net breakage terms. The integrals in the aggregation term were estimated using a Gaussian-Laguerre quadrature with 9 terms. After the integration, the MSM generated the calculated PSD profiles (between the boundary conditions, inclusive) for all four data sets used. The optimization routine extracted the estimated mean particle sizes and the CVs representing the interior ports (2 and 3) for substitution into the objective function. The method is repeated until the objective function is satisfied and the consecutive values of the calculated parameters satisfy the set tolerance (10^-4).

The set of the optimum parameters obtained from this study are presented in Table 5 and the model fits to the experimental data are presented in Figure 9A. Figure 10 shows the calculated profiles for the nucleation and growth rates along the tubular precipitator. Both rates decrease progressively along the tubular precipitator as the degree of supersaturation decreases. This confirms the observed increase in solids concentration (Figure 5) with residence time, which was due to the formation of new solids from the solution. Therefore, the assumption that the entire solid protein comes out of the solution in 1 s is not realistic.

Figure 10 — Calculated nucleation and growth rates along the tubular precipitator using optimized parameters.

MODEL PREDICTIONS OF THE EXPERIMENTAL RESULTS

The kinetic parameters obtained from the experimental data were used to predict the PSD from four different experiments (Table 5). Due to the lack of experimental data on the kinetics of the sunflower protein precipitation in the literature, the validation of the model was performed using our own data. This time, instead of solving the PBE as a boundary value problem, it was solved as an IVP (initial value problem) using the LSODE algorithm (Hindmarsh, 1980). In order to start the integration, it was required to supply the initial PSD. It was hoped that, by supplying the feed concentration and the total flow velocity, the program should be able to generate the entire PSD profile for the tubular precipitator (starting from initial nucleation at ζ = 0). It turned out that, the problem was very stiff and at reduced tolerances, the predicted values showed a large discrepancy from
the experimental values. To circumvent this stiffness, the experimental data from the first port \( (\zeta = 0.02, \text{ close to the precipitator entrance}) \) were used as the initial condition. Predicted and experimental results are compared in Figure 9B. For small mean particle sizes, the predicted results showed a wider deviation from the experimental results. When the experimental errors are taken into account the discrepancies are within the experimental errors.

**Critical Evaluation of the Kinetics of Precipitation and Kinetic Parameters**

The models presented here for nucleation, growth, breakage, and aggregation rates, although not fundamental, are based on sound assumptions and represent logical dependence on key process variables. For example, the primary nucleation rate is shown in Equation (8c) to be a strong function of supersaturation. The supersaturation decreases along the length of the tubular precipitator and therefore, nucleation rate is correlated with the dimensionless reactor length. A similar dependence on the supersaturation and the dimensionless reactor length is assumed for the growth rate in Equation (9c). Equation (10) demonstrates a size-dependent death rate due to breakage. The larger the particles, the higher is the probability of their breakage upon collision. Since particle breakage is a volume-conserved process, Equation (11) associates the birth rate of particle of size \( L \) to the death rate of particles of size \( 2^{1/3}L \). The aggregation rate is represented by Equation (15), which is based on Smoluchowski’s model with a size-independent aggregation kernel. There is no attempt in the literature to model such a complex process involving nucleation, growth, breakage, and aggregation sub-processes in a tubular reactor. The overall model and its solution methodology, presented in the present work, are the first attempt to undertake such a task. The overall model involves too many parameters (11 in total). This, however, is inevitable due to the lack of fundamental theory on the nature of nucleation, growth, breakage, and aggregation sub-processes. Another limiting fact is the lack of previous experimental data in the literature on these sub-processes involving the precipitation of proteins. As more fundamental models for nucleation, growth, aggregation, and breakage sub-processes become available in the future, the limitation of the overall model will be relaxed and the number of estimated kinetic parameters will be reduced. In the present work, we have shown that the model, despite these shortcomings, predicts correct qualitative trends (Figures 6 to 8) and particle size distribution (Figure 9B) using the optimized kinetic parameters.

**Conclusions**

In this study, kinetics parameters of the sunflower protein precipitation in the tubular precipitator by isoelectric method (aqueous HCl) were determined. The non-linear PBE of the tubular precipitator including primary nucleation, molecular growth, growth by aggregation, and breakage of aggregates was solved using the orthogonal-collocation and multiple shooting methods. The estimated kinetics parameters were used to check the validity of the model by predicting the PSD of experiments which were not used in the parameter estimation algorithm (Figure 9B). The predicted PSDs along the tubular precipitator were within the experimental errors. Using these kinetic parameters, the length of the tubular precipitator required to give the desired mean particle size of the precipitate can be calculated.

**Appendix: Solution of the PBE**

Equation (16) was used to approximate both interior and initial population densities. This method is referred to as mixed method. Equation (16) has a total of \( n \) adjustable parameters, \( \alpha_i(\zeta) \). One for the initial condition \( (\zeta = 0) \) and \( (n-1) \) for the interior points. Equation 16 can be written in matrix form as:

\[
\bar{\mathbf{Y}} = \Phi \alpha \quad \text{...........................................} \quad (A.1)
\]

where:

\[
\alpha = [\alpha_0(\zeta) \quad \alpha_1(\zeta) \quad \alpha_2(\zeta) \quad \ldots \quad \alpha_{n-1}(\zeta)]^T \quad \text{...........................................} \quad (A.2)
\]

\[
\bar{\mathbf{Y}} = [y(x_1,\zeta) \quad y(x_2,\zeta) \quad \ldots \quad y(x_n,\zeta)]^T \quad \text{...........................................} \quad (A.3)
\]

and,

\[
\Phi = \begin{bmatrix}
\phi_{1,0} & \phi_{1,1} & \ldots & \phi_{1,n-1} \\
\phi_{2,0} & \phi_{2,1} & \ldots & \phi_{2,n-1} \\
\vdots & \vdots & \ddots & \vdots \\
\phi_{n,0} & \phi_{n,1} & \ldots & \phi_{n,n-1}
\end{bmatrix} \quad \text{...........................................} \quad (A.4)
\]

where \( \phi_{ik} = e^{-x_j} [\psi_i(x_k) - \psi_j(x_k)] \) for a given \( x = x_i; (i = 1, 2, 3, \ldots, n) \) also \( \psi_0 = 0 \) and \( \psi_1 = 1 \). \( x_r \) are the \( n \) roots of the Laguerre polynomial.

**Particle Aggregation**

The integrals in the aggregation equation, Equation (15), were solved using Gaussian-Laguerre quadrature defined as:

\[
\int_0^\infty h(x)dx = \sum_{i=1}^n w_i e^{x_i^*} h(x_i) \quad \text{...........................................} \quad (A.11)
\]
where \( x_i \) are the \( n \) zeros of the Laguerre polynomial and \( w_i \) are the corresponding weighting factors. Equation (15) can be written in matrix form as:

\[
\begin{align*}
\mathbf{J} &= \mathbf{P} \begin{bmatrix} \frac{x_1}{2} & I_{x_1} \end{bmatrix} \mathbf{K} \begin{bmatrix} I_{x_1} \end{bmatrix} \mathbf{Y} \end{align*}
\]

(12.12)

where:

\[
\mathbf{A} = \begin{bmatrix} A(x_1, \zeta) & A(x_2, \zeta) & A(x_3, \zeta) & \ldots & A(x_n, \zeta) \end{bmatrix}^T
\]

(13.13)

\[
\mathbf{K} = \begin{bmatrix} K_{x_1} & K_{x_2} & \ldots & K_{x_n} \end{bmatrix}^T
\]

(14.14)

and,

\[
I_{x_k} = \int_0^{x_k} f(x, \zeta) \, dx = \sum_{i=1}^{n} w_i e^{x_i} f(x_i, \zeta)
\]

(15.15)

\[
y(q_k, \zeta) = \frac{y(q_k, \zeta)}{q_k^{3/2}}
\]

(16.16)

\[
y(q_k, \zeta) = \sum_{i=k}^{n} \alpha_i e^{-q_k \psi_{j-1}(q_k)}
\]

(17.17)

Therefore,

\[
\begin{align*}
I_{x_1} &= w_1 e^{x_1} f(x_1, \zeta) \\
I_{x_2} &= I_{x_1} + w_2 e^{x_2} f(x_2, \zeta) \\
I_{x_n} &= I_{x_{n-1}} + w_n e^{x_n} f(x_n, \zeta)
\end{align*}
\]

(18.18)

The term \( I_x \) is given by:

\[
I_x = \int_0^{x} y(x, \zeta) \, dx = \sum_{i=1}^{n} w_i e^{x_i} y(x_i, \zeta)
\]

(19.19)

Since \( c_i \) are values of \( x \leq x_n \), then in Equation (19.19) \( c_i \) are replaced by \( x_i \).

**Breakage rate**

The breakage rate, Equation (13.13), can be represented in matrix form as:

\[
\mathbf{B} = \mathbf{B}_{\text{on}} \begin{bmatrix} \bar{Y}_1 - \bar{Y}_2 \end{bmatrix}
\]

(20.20)

where \( \mathbf{B}_{\text{on}} = k_j N_j L_j \begin{bmatrix} d \end{bmatrix}^{-1} \) and:

\[
\begin{align*}
\bar{Y}_1 &= \begin{bmatrix} y(x_1, \zeta) & y(x_2, \zeta) & \ldots & y(x_n, \zeta) \end{bmatrix}^T \\
\bar{Y}_2 &= \begin{bmatrix} x_1^1 & x_1^2 & \ldots & x_n^2 \end{bmatrix}^T
\end{align*}
\]

(21.21)

**Node by node calculations**

The node points values of \( \alpha_i(\zeta) \) at \( x_i, i = 1, 2, \ldots, n \) and \( \zeta_j, j = 1, 2, \ldots, m \) are represented Equation (19). The second order ordinary differential equations can be transformed into a set of first order ordinary differential equations as:

\[
\frac{d}{d\zeta} \begin{bmatrix} \alpha_1 \\ \alpha_2 \\ \vdots \\ \alpha_n \end{bmatrix} = \begin{bmatrix} \mathbf{f}_1 \\ \mathbf{f}_2 \\ \vdots \\ \mathbf{f}_n \end{bmatrix}
\]

(26a)

\[
\begin{align*}
\mathbf{f}_1 &= \begin{bmatrix} \frac{J}{N_0} \end{bmatrix} \\
\mathbf{f}_2 &= \begin{bmatrix} \frac{J}{N_0} \end{bmatrix} + \begin{bmatrix} \frac{J}{N_0} \end{bmatrix} \\
&= \begin{bmatrix} \frac{J}{N_0} \end{bmatrix} - \begin{bmatrix} \frac{J}{N_0} \end{bmatrix}
\end{align*}
\]

(26b)

Therefore, a total of 17 first order ODE are to be solved (16 equations represent the interior nodes and 1 equation represents the initial condition). However, in order to solve Equation (26a) we need to define the equation defined by \( \alpha_i \), which is given by the IC (initial condition), so that we have a closed set of equations (17 equations and 17 unknowns; 8 \( \alpha_i \)s and 9 \( \zeta \)s). Substituting the approximate solution into the initial condition equation, Equation (20.1a), and taking the limit as the size \( x_i \rightarrow 0 \) gives:

\[
\sum_{i=1}^{n} \alpha_i(\zeta) \psi_{j-1}(x_k)_{x_i \rightarrow 0} - \begin{bmatrix} \frac{J}{N_0} \end{bmatrix} = R_k(0, \zeta), x \rightarrow 0(0 \leq \zeta < 1)
\]

(27)

Differentiating once Equation (27) with respect to \( \zeta \) gives:

\[
\frac{d}{d\zeta} \begin{bmatrix} \alpha_1(0) + \alpha_2(0) + \alpha_3(0) + \ldots + \alpha_n(0) \end{bmatrix} = 0
\]

(28)
where:
\[
\frac{d}{dz} \left[ \frac{J / \dot{N}_o}{G / L_c} \right]_{z=0} = \frac{J_0 L_c (1-k_{BO} \zeta)^2}{G_0 N_0 (1-k_{BO} \zeta)^3} \left[ \frac{b_k g}{1-k_{BO} \zeta} - \frac{a k_{BO}}{1-k_{BO} \zeta} \right]
\]

(A.29)
solving for \( \alpha' \) and substituting for \( f \) results in:
\[
\alpha' = \frac{1}{\psi_0(0)} \left[ \frac{J L_c (1-k_{BO} \zeta)^2}{G_0 N_0 (1-k_{BO} \zeta)^3} \left[ \frac{b_k g}{1-k_{BO} \zeta} - \frac{a k_{BO}}{1-k_{BO} \zeta} \right] \right]
\]

(A.30)
where \( \psi_0(0) = 1 \). Therefore, now we have a closed system of equations. This set of 17 equations is solved at different values of \( \zeta \) in this study. The number of node points along the \( \zeta \) direction was 101. This means a total of 101 x 17 first order ODEs are solved each time. The corresponding boundary condition at the entrance, \( \zeta = 0, \psi(z) \rightarrow \infty \) (approaches mixed flow). Therefore the BC at the entrance can be simplified as:
\[
\alpha = 0; \quad \zeta = 0 \quad (x > 0) \quad \ldots \quad (A.31)
\]
or if at any other location, then the boundary conditions corresponds to the PSD at that location. The boundary condition at the exit \( \zeta = 1 \) is given by:
\[
\tilde{f} = 0; \quad \zeta = 0 \quad (x > 0) \quad \ldots \quad (A.32)
\]
which means \( \alpha' \) 's are constant at the exit. For known PSD at the exit, \( Y(x,1) \), Equation (20c) can be integrated once to give:
\[
\alpha = \frac{-1}{L} \tilde{Y}; \quad \zeta = 1 \quad (x > 0) \quad \ldots \quad (A.33)
\]
This form of boundary condition was used during determination of kinetic parameters from the experimental data. In that case the values of \( Y(x,\zeta) \) at the four sampling ports (different \( \zeta \)) are known.

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**Nomenclature**

- \( a \) = nucleation parameter
- \( A \) = net rate of population density change due to particle aggregation, (no/\( \mu \)mL/s)
- \( A_T \) = total solids surface area per volume of solid free liquid, (m²/m³)
- \( b \) = growth rate parameter
- \( B \) = net rate of population density change due to particle breakage, (no/\( \mu \)mL/s)
- \( B_i \) = particle birth rate due to aggregation, (no/\( \mu \)mL/s)
- \( B_{i,0} \) = particle birth rate due to breakage, (no/\( \mu \)mL/s)
- \( B_{con} \) = parameter defined in Equation (A.20)
- \( C \) = protein concentration, (kg/mL of solid free liquid)
- \( C_0 \) = protein concentration in the feed stream, (kg/mL³)
- \( C^* \) = equilibrium protein concentration, (kg/mL³)

- \( CV \) = experimental coefficient of variation based on volume distribution
- \( CV_{i,j} \) = coefficient of variation based on volume distribution for run \( i \) at jth port
- \( CV_{m,i,j} \) = calculated coefficient of variation based on volume distribution for run \( i \) at jth port
- \( d \) = breakage parameter, Equation (10)
- \( d_m \) = mean particle size for run \( i \) at jth port, (\( \mu \)m³/\( \mu \)m³)
- \( d_m \) = calculated mean particle size, (\( \mu \)m³/\( \mu \)m³)
- \( d \) = diameter of the tubular precipitator, (cm)
- \( D_a \) = particle death rate due to aggregation, (no/\( \mu \)mL/s)
- \( D_m \) = particle death rate due to breakage, (no/\( \mu \)mL/s)
- \( D_t \) = axial dispersion coefficient, (cm²/s)
- \( f \) = Fanning friction factor
- \( f_n(\zeta) \) = function defined in Equation (A.16)
- \( f_{i,j} \) = frequency of orthokinetic collisions, Equation (1a)
- \( g \) = growth rate parameter, Equation (9a)
- \( G \) = particle growth rate, (cm/s)
- \( G_p \) = aggregate growth rate, (cm/s)
- \( G_{p,i} \) = initial particle growth rate, (cm/s)
- \( H \) = objective equation to be minimized
- \( IC \) = initial condition in Equation (A.26)
- \( I_{x_k} \) = function defined in Equation (A.18)
- \( I_{x_k} \) = function defined in Equation (A.19)
- \( J \) = nucleation rate, (no/mL/s)
- \( J_i \) = initial nucleation rate, (no/mL/s)
- \( k_h \) = specific area shape factor
- \( k_n \) = nucleation parameter, Equation (8a)
- \( k_{n,i} \) = nucleation parameter, Equation (8a)
- \( k_{n,i} \) = nucleation parameter, Equation (8a)
- \( k_i \) = growth rate parameter, Equation (9a)
- \( k_{i,i} \) = growth rate parameter, Equation (9a)
- \( k_{i,i} \) = growth rate parameter, Equation (9a)
- \( k_{i,i} \) = model parameter, Equation (22)
- \( k_{i,i} \) = model parameter, Equation (22)
- \( k_i \) = location of measuring point along the precipitator, (cm or m)
- \( L \) = particle or aggregate size, (\( \mu \)m)
- \( L_i \) = particle size scaling factor, (3.528 \( \mu \)m)
- \( L_i \) = particle or aggregate size at the ith channel or interval, (\( \mu \)m)
- \( L_{n} \) = nucleic particle size, (\( \mu \)m)
- \( M_{T} \) = solid protein concentration, (kg/mL of solid free liquid)
- \( n \) = number of node points used in approximation, Equation (16)
- \( N \) = particle concentration, (no/mL)
- \( N_{i} \) = particle concentration in the size range i, (no/mL)
- \( N_{i} \) = population density scaling factor, (10³ no/mL)
- \( p \) = population density, (no/\( \mu \)mL/s)
- \( p \) = population density, (no/\( \mu \)mL/s)
- \( P \) = matrix of parameters to be optimized, Equation (23)
- \( P_{i} \) = constant in Equations (15), (20) and (A.12)
- \( Pe \) = Peclet number, (L/D)
- \( Q \) = total liquid feed flow rate, (m³/s)
- \( R_i \) = volumetric feed ratio, (precipitant/protein solution)
- \( r \) = mean radius of particles in the size range i, (\( \mu \)m)
- \( R_i \) = residual matrix, Equation (19)
- \( R_{T} \) = aggregation rate, (no/\( \mu \)mL/s)
- \( R_{i} \) = aggregation rate, (no/\( \mu \)mL/s)
- \( R_{i,0} \) = aggregation rate, (no/\( \mu \)mL/s)
- \( S_i \) = supersaturation ratio
- \( S_{i,0} \) = mean residence time, (s)
- \( T \) = matrix defined by Equation (A.10)
- \( u \) = superficial liquid velocity, (cm/s)
- \( u_{i} \) = superficial liquid velocity, (cm/s)
- \( v_i \) = total volume fraction of particles at the ith channel or interval
- \( V \) = precipitator volume, (mL)
- \( W \) = weighting factors of Gaussian-Laguerre quadrature, Equation (A.11)
- \( W_i \) = weighting factors of Gaussian-Laguerre quadrature, Equation (A.11)
- \( W \) = solid protein concentrations, (kg/mL of solid free liquid)
\( x \) = dimensionless length along the tubular precipitator, Equation (5)
\( x_i \) = \( n \) zeros of the Laguerre polynomial
\( y \) = dimensionless population density, Equation (5)
\( Y \) = matrix of \( y \), Equation (A.2)
\( z \) = precipitator length, (cm, m)
\( z_i \) = total length of the precipitator, (cm, m)

\[ \gamma \]

Greek letters
\( \alpha \) = weighting parameter in Equation (16)
\( \alpha \) = matrix of \( \alpha \), Equation (A.3)
\( \beta \) = aggregation kernel in Equation (14), (mL/s)
\( \beta_0 \) = constant aggregation kernel, (mL/s)
\( \gamma \) = elements of matrix, Equation (33)
\( \delta \) = Kronecker delta, (0 if \( \gamma \neq \gamma \))
\( \delta_{ij} \) = dimensionless particle size
\( e \) = dimensionless particle size less than \( x \), Equation (A.15)
\( \eta \) = rate of energy dissipation per unit mass, (W/kg)
\( \lambda_{ij} \) = variable in Equations (1b) and (1c)
\( \mu_a \) = model parameter in Equation (22)
\( \mu_s \) = model parameter in Equation (21)
\( \nu \) = kinematic viscosity of liquid, (m^2/s)
\( \nu_r \) = kinematic viscosity of liquid, (m^2/s)
\( \sigma \) = apparent relative supersaturation
\( \zeta \) = mean residence time of solid-free liquid, (s)
\( \Phi \) = function defined in Equation (17)
\( \Phi \) = matrix defined in Equation (A.4)
\( \Phi \) = elements of matrix Equation (A.4)
\( \psi_j \) = \( j \)th term in Laguerre polynomials, Equation (17)

Abbreviations
MSM = multiple shooting method
MWR = method of weighted residue
OCM = orthogonal collocation method
PBE = population balance equation
PSD = particle size distribution
SAM = simulated annealing method

References

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